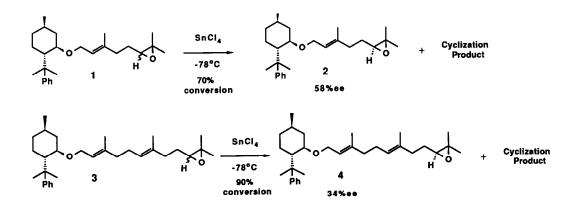
A NEW SYNTHETIC ROUTE TO JUVENILE HORMONE KINETIC RESOLUTION OF EPOXIDES USING ORGANOALUMINUM REAGENT

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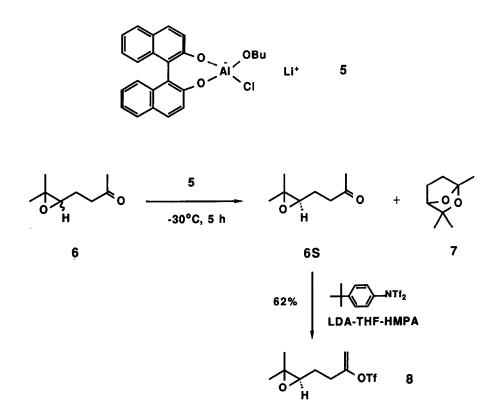
<u>ABSTRACT</u>: A short synthetic route to C_{16} -juvenile hormone is described which depends on new oxirane ring opening methodology and which also involves the joining of intermediate 8 and 12 in a one flask operation to construct the sesquiterpene structure.

This note describes a new synthetic route to C_{16} -<u>Cecropia</u>-juvenile hormone (14)¹ in optically active form. The synthesis is based on unique aspects of the chemistry of keto-epoxides and features the assembling of C_7 - and C_8 -segments of acyclic sesquiterpene in single process.

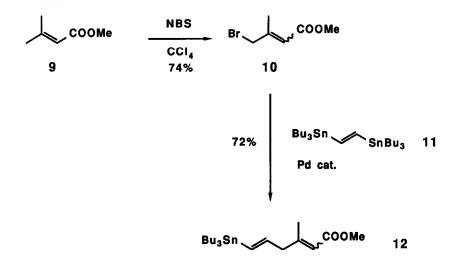
Past studies have revealed that the chiral anchimeric assistance in the epoxide ring opening process is possible and should provide a new route to optically active epoxide with the remote asymmetric induction, e.g. $1 \rightarrow 2^2$ and $3 \rightarrow 4$.³ Thus, a high degree of chiral neighboring π -bond participation during S_N^2 -like epoxide ring opening should play an important role for the formation of A ring during the biological construction of steroid.⁴ In parallel with these studies we have investigated an approach to the chiral Lewis acid catalyzed ring opening of epoxides in systems where a similar anchimeric assistance in the epoxide ring opening process can be expected.



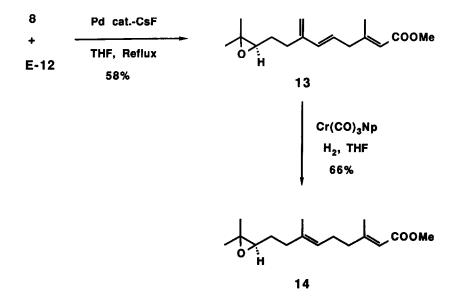
Chiral Lewis acid 5^5 was synthesized in situ from the R(+)-binaphthol, dimethylaluminum chloride, and lithium butoxide in dichloromethane. Use of this catalyst to cyclize the polyene terminal epoxides did not proceed smoothly at low temperature, and at the higher temperature the effective kinetic resolution could not be achieved. The aim of the present study was to see if we could enhance the rate of the ring opening by use of an appropriately placed cation Accordingly, we elect to compare the cyclization of 6 having the carbonyl auxiliary stabilizer. at C-6. Indeed, an efficient but unconventional process for the separation of the enantiomers 6 was thus developed: treatment of the racemate 6 with the freshly prepared 5 at -30° C for 5 h to give the optically pure epoxide $R-6^6$ in addition to the internal acetal 7. This process is convenient and amenable to scale-up. The optically pure R-isomer can be obtained in ca. 20% yield from the readily available racemate 6; higher yields are probably achievable. Recovery of binaphthol is excellent (>90%). The pure epoxide R-6, which was now shown to be a highly useful building block in acyclic terpene synthesis, was converted to the triflate 8 with triflic imide⁷ and lithium diisopropylamide in THF- HMPA at low temperature.



The other component required for the assembly of juvenile hormone was obtained in a straightforward manner. Radical bromination of the ester 9 with <u>N</u>-bromosuccimide in carbon tetrachloride gave rise to the bromoester 10 ($\underline{E}/\underline{Z} = \underline{ca.} 1:1$),⁸ which was coupled with the ditin 11 in the presence bis(acetonitrile)dichloropalladium using the method of Stille⁹ to afford the ester 12 in 72% yield. After careful chromatographic separation of the stereoisomer, the pure E-12 was used for the subsequent coupling reaction.



The JH system was now assembled from components 8 and 12 as follows. A mixture of tetrakis(triphenylphosphine)palladium (2 mol%), cesium floride (3.0 equiv),¹⁰ 8, and 12 in dry degassed THF was heated at reflux for 5 h, after column chromatography on silica gel, to produce the desired triene 13 in 58% yield.¹¹ Partial and selective hydrogenation of 13 was accomplished in the presence of tricarbonyl[(1,2,3,4,4a,8a- η)naphthalene]chrominum catalyst¹² under 80 atm of H₂ in dry THF at 45^oC for 5 h to afford the juvenile hormone 14 in 62% yield without contamination of any of the corresponding \underline{Z} isomer.¹³



The above reported synthesis of juvenile hormone contains a number of noteworthy methodological elements including (1) the novel synthesis of optically active epoxide 6, (2) the one flask process for the assembling of right and left hand side of sesquiterpene structure, (3) stereospecific construction of trisubstituted olefin using chrominum catalyzed hydrogenation of 1,3-

Acknowledgment. This research was supported by Grant-in Aids from the Ministry of Education, Science and Culture, Japan. One of us (YN) was also acknowledged for the JSPS Fellowships for Japanese Junior Scientists.

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- 6. $[\alpha]_D^{21} = 28.3^\circ$ (c = 0.94, CHCl₃). The optical purity was determined with ¹H NMR analysis in the presence of Eu(hfc)₃, see ref. 2.
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 Stille, J. K., ibid., <u>1986</u>, 108, 3033; Stille, J. K.; Tanaka, M., ibid., <u>1987</u>, 109, 3785.
- 11. IR (neat film) 3200-2800s, 1735s, 1660s, 1445m, 1230w; ¹H NMR (CDCl₃) δ 6.14 (d, 14 Hz, 1H), 5.70 (s, 1H), 5.66 (dd, 14 and 8.5 Hz, 1H), 5.00 (d, 2 Hz, 1H), 4.98 (d, 2 Hz, 1H), 3.70 (s, 3H), 2.93 (d, 8.5 Hz, 2H), 2.76 (t, 6 Hz, 1H), 2.42 (m, 1H), 2.31 (m, 1H), 2.16 (s, 3H), 1.73 (m, 2H), 1.31 (s, 3H), 1.26 (s, 3H); Anal. Calcd for C₁₆H₂₄O₃: C, 72.7; H, 9.2. Found: C, 72.4; H, 9.1. [α]_D²¹ = 7.6^o (c 0.94, ether).
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- 13. IR (neat film) 3200-2800s, 1735s, 1660s, 1445m, 1235w, 1160s; ¹H NMR (CDCl₃) δ 5.68 (br. s, 1H), 5.15 (br. s, 1H), 3.69 (s, 3H), 2.71 (t, 6.8 Hz, 1H), 2.19 (s, 3H), 2.18 (s, 3H), 2.11 (m, 2H), 1.7-1.5 (m, 6H), 1.31 (s, 3H), 1.26 (s, 3H); Anal. Calcd for C₁₆H₂₆O₃: C, 72.1, H, 9.8. Found: C, 71.9, H, 9.6; $[\alpha]_D^{21} = 5.4^\circ$ (c = 0.6, methanol), lit. $[\alpha]_D = 5.75^\circ$ (c = 0.4, methanol) for (R)-isomer, and $[\alpha]_D = -5.44^\circ$ (c = 0.7, methanol) for (S)-isomer, see ref. 1.

(Received in Japan 28 December 1987)

diene.